

Thermal spectra now well-explained!

- quanta of energy key / discretization
- oscillators only emit "packets" of $E = h\nu = hf$
- correct classical result when $h \rightarrow 0$!!!

Problem? non signy or atomic emission/absorption

2 ways to view interaction of light & sparse atomic matter

(1) sparse gas, send white light in. see what makes it through

\Rightarrow Discrete absorption lines!

- even w/ "quanta" we got a cont. thermal spec
- we miss something yet

(2) excite sparse gas w/ high voltage

\Rightarrow discrete emission, same λ as absorb!
(few missing)
same process!

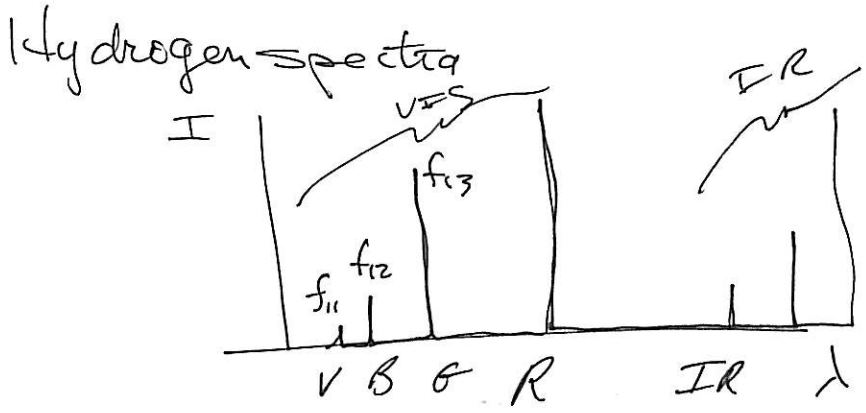
(*) NOT thermal

NOT acc chg - eg. cyclotron - continuous!

Further: stability of atoms still in question!
cannot expl stable neutral matter...

- This implies something key: somehow atomic energy levels have discrete too, not just emitted rad!
- discreteness goes beyond Planck's assumption
- Will come back to this in Sect III; for now a sketch

- emission of light as a vibration of sorts?
 - if so, frequency of vibr \rightarrow rad freq; harmonics
 - already discounted this



- discrete lines
- char. series that look & in UV series, Vis, IR, ...

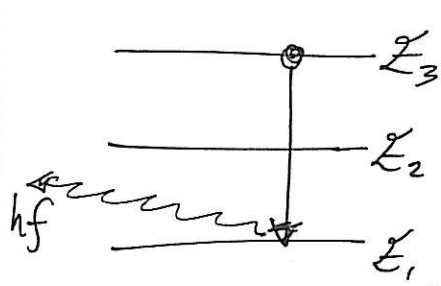
can index frequencies of emission

e.g. $f_{11} \quad f_{12} \quad \dots \quad f_{21}, f_{22}, \dots$

Curious: any sum of 2 freq's gives a 3rd observed!

- Strongly implies that allowed energy states of atoms are discrete, not just rad!
- So why is the discharge spectra continuous in intensity i.e. no flicker? MANY atoms!

Bohr had an "expl" - take plank further, atoms only emit in certain discrete states



light emitted in quanta when atom transitions from high \rightarrow low state

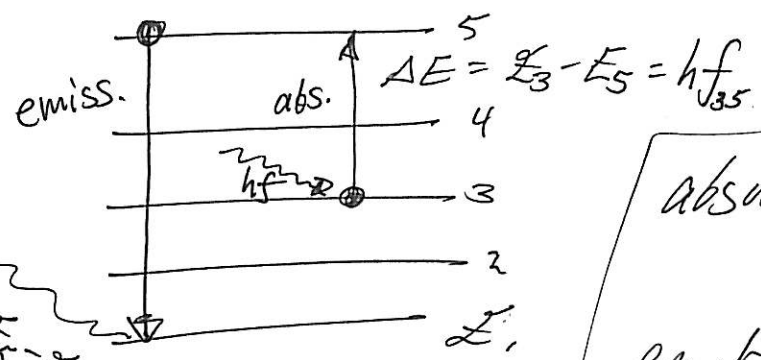
E_{emitted} :

$$E_3 = E_1 + h\nu$$

Rad discrete
 ..
 levels discrete
 (tautology)

- if $h\nu$ is discrete ... then so must $E_3 - E_1$ or any $E_i - E_j$!
- can't have cont. energy of atom \Rightarrow discrete light!

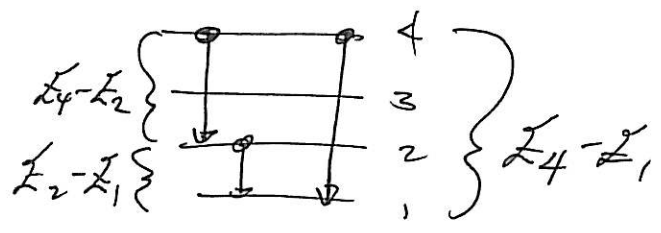
Bohr: Planck's bundles are discrete trans of atoms. E levels transmit: absorb hf of light = change in level
emission = trans \downarrow , emit hf to go to lower level



absorb: ~~$E_{\text{init}} + hf = E_f$~~
 $E_{\text{init}} + hf = E_f$
 emit: $E_{\text{init}} = E_f + hf$

$$\Delta E = hf = E_5 - E_1$$

explains sum rule



$$hf_{42} = E_4 - E_2$$

$$hf_{21} = E_2 - E_1$$

$$hf_{41} = E_4 - E_1 = (E_4 - E_2) + (E_2 - E_1) = hf_{42} + hf_{21}$$

$$\rightarrow f_{41} = f_{42} + f_{21} = \frac{E_4 - E_1}{h}$$

$$f_{nl} = f_{nm} + f_{ml}$$

Boltz factor expl why this isn't constantly happening
 and why there are often more emission than absorption lines

• can't **absorb** radiation to go from 3 to 4 unless its already in 3!

• prob of being in state 3?

$$\frac{N_3}{N_1} = \frac{e^{-E_3/kT}}{e^{-E_1/kT}} = e^{-(E_3 - E_1)/kT}$$

$$\sim e^{-30} \dots \text{no way.}$$

$$\Delta E \sim 2 \text{ eV}$$

at RT, nearly all atoms @ lowest level!

- at RT, expect all emissions possible if excited electrically
- only absorptions from 1 → higher are likely

very high temps: start to see other abs.

- 2 → higher
- 3 → higher etc

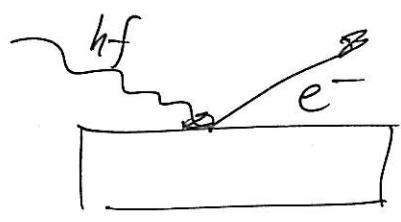
Much more on this in Sect III

- main point: making rad discrete has CONSEQ
- atom energies also discrete ... what else?

Next what other empirical prop point toward new phys?

photoelectric effect

basically: shine light on a metal surface
electrons are ejected



- light has momentum, etc
- consistent w/ wave beh.

problems? in the details.

if light = waves, how is energy discrete?
how can it be discrete if emitted?

Planck's energy quanta were a formality - not taken literally!

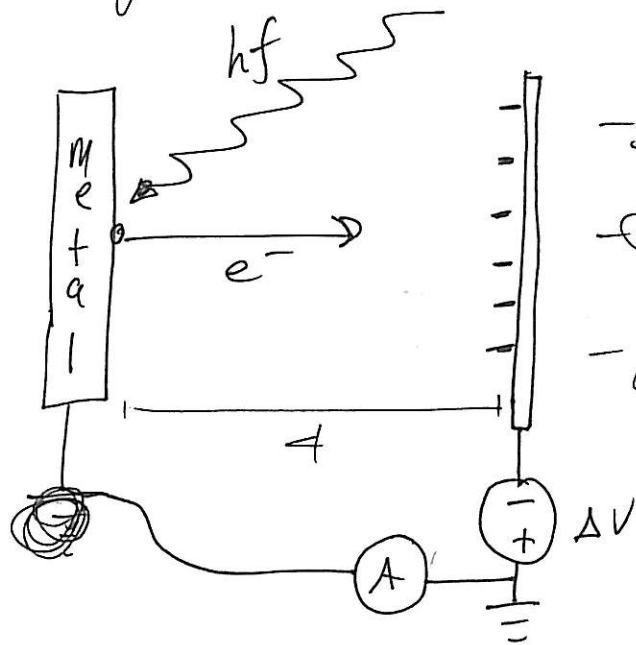
- light still viewed as a wave
... so how is energy discrete?

- deeply confusing times !!

Einstein - take Planck literally!

- if energy is discrete, then so is light itself!

- why? the exp't



- light ejects e^- (ONLY! emp.)

- ~~ΔV accelerate~~

- absorbing light via KE

- NEG metal plate

if $KE < e\Delta V$, NO I

• ~~atp~~ • checks E distr of electrons, cf light freq!

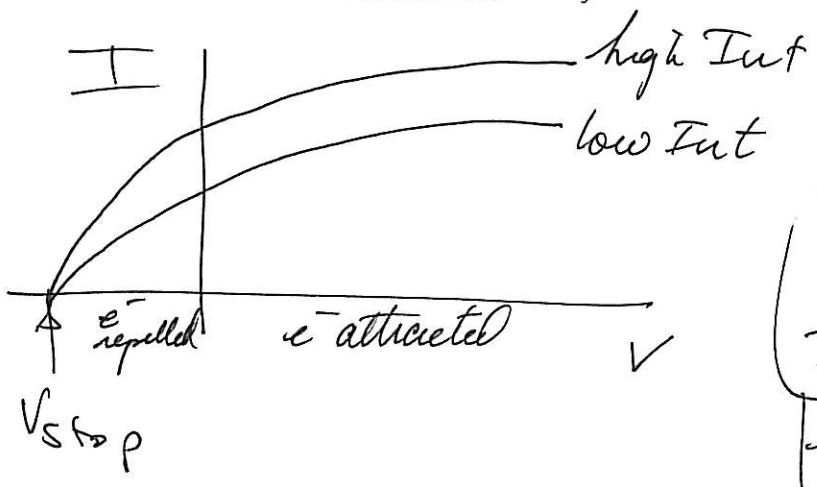
$$K_{e^-} = \frac{1}{2}mv^2 = hf \quad \text{photon} \rightarrow \text{electron KE}$$

if $\frac{1}{2}mv^2 \geq e\Delta V$ e^- collected, current!

ΔV Stop measures e^- KE!

vary ΔV for given f ; at some cut $\Delta V < 0$

no e^- make it any more - STOPPING POTENTIAL



problem?

Stopping potential
 INDEP of intensity!
 higher intensity should
 mean more energy?

Conclusion: Light energy is NOT related to intensity!
 energy always SAME given f

if it were, expect $V_{stop} \sim I_{intensity}$

Classical expect

Reality

e^- absorb EM waves energy continuously
~~current~~
 } so intensity means KE \uparrow means higher ΔV_{stop}

KE_{max} indep of intensity

time delay - turn light on,
 wait for e^- to absorb enough E
 varies w/ intensity

NIKS. $\ll 10^{-9}$ sec

e^- ejected @ any f so long
as intensity high enough

e^- only come out if
 $f > \text{crit. value!}$

KE and f unrelated

KE \uparrow as $f \uparrow$

major FAIL!

Einstein's model: incident light is quanta!

- light emitted/absorbed as quanta AND travels as such!
- light = bundles, massless particles! photons

- $\frac{1}{r^2}$ intensity drop OK ... particles spread out

- if $m=0$, travel at $v=c$ in straight lines

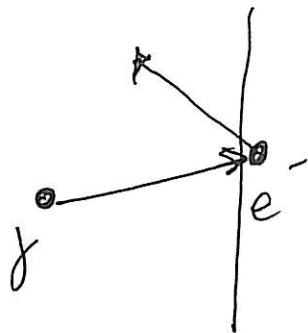
- energy \neq intensity

intensity = more particles of same energy each

$E_{ph} = hf$ as Planck said

now its a mechanics problem!

γ = photon



relativity: if $m=0$, $\vec{p} = \frac{E}{c} = \frac{hf}{c} = \frac{h}{\lambda}$

= photons still have momentum

- they hit e^- , eject them

if their energy $>$ Binding E

Simplest: light quanta (photon) gives all energy to e^- (\vec{p} too). if its enough, e^- ejected

- electrons are bound to crystal
- mostly e^- near surface (no scatter/recapture)
- mostly ejected w/ $\vec{v} \perp$ surface (\vec{p} consv.)

E balance:

$$\textcircled{C} \quad \textcircled{D}$$
$$E_{\text{photon}} = KE_{e^-} + BE_e = KE + W$$

$\underbrace{\hspace{10em}}_{hf} \qquad \underbrace{\hspace{10em}}_{\text{binding energy aka work fn}}$

or, $K_{e^-} = hf - W$

this explains it all!

• have to have $hf > W$ for any ejection!
exceed binding energy of e^- to crystal

• KE linear in freq above that

• instantly - each photon gives all its energy
one photon \rightarrow one e^-

• Intensity is irrelevant, has nothing to do w/ E
classical optics: not true... but not in ems.

$$E_{\text{light}} = (\# \text{ photons}) (\text{energy per photon}) \\ = N hf$$

$$\text{intensity} \sim N/\Delta t$$

• weak beam does the same, just less photons/sec!

• ~~Since~~ Since W depends on material, so does
freq required for e^- emission ✓

• why film less sensitive to red light... less $E_{\text{per photon}}$!

? what more does this imply

? why do ray diagrams work AND wave interference?

Scale of probe

releu dist $\sim \lambda$

wave-like



releu dist $\gg \lambda$

ray/particle-like

wave packet



really, neither!!! just our lack of imagination. MORE TO COME

Impl from relativity

$E_{ph} = hf = h\nu$ Planck-Einstein

$E^2 = m^2 c^4 + p^2 c^2$ Relativity
 $v_{light} = c$

- Only consistent if $m_{\text{photon}} = 0$!

= implies $E_{ph}^2 = c^2 p^2$ or $|\vec{p}| = \frac{E_{ph}}{c}$

or $|\vec{p}| = \frac{hf}{c} = \frac{h}{\lambda}$

- So, wacky: momentum w/o mass by virtue of energy
- however: emission of photon means a body loses energy E_{ph} , requiring its mass to decrease by E/c^2 !
- absorption photon: mass-energy eqn \Rightarrow mass increase
 - * tiny * though. green photon $\sim 2\text{eV}$
 - electron mass: $511,000\text{eV}$
- also: radiation pressure (also classically)
consistent w/ classical E \dot{S} M

? How to prove photons have momentum
Collide them with stuff.

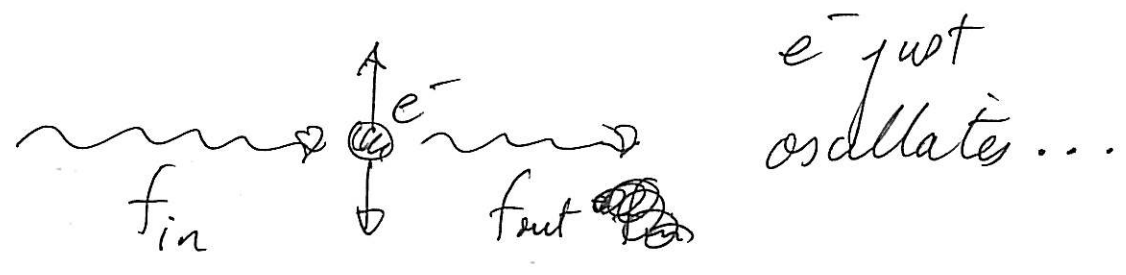
COMPTON EFFECT

experiment: shoot high energy (gamma or x) light at e^-

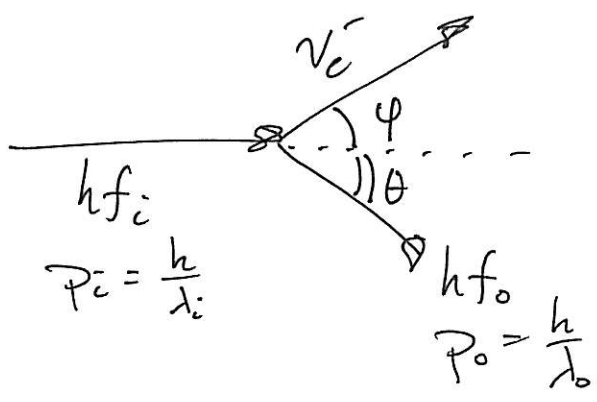


- photon ejects e^- ... scattered
 - if e^- acquires KE, photon **LOSES** energy!
 - $f_{in} > f_{out}$

Classical



- result "proves" photon model has **intrinsic momentum + particle-like scattering**
- high energy \rightarrow short λ \rightarrow particle-like
 (low energy \rightarrow long λ \rightarrow wave-like)
 depending on relevant scale



- conserve \vec{p}, E
- use relativistic E formulas (see text)

$$\Rightarrow \boxed{\lambda_o - \lambda_i = \left(\frac{h}{mc}\right)(1 - \cos\theta)} = \lambda_c (1 - \cos\theta)$$

$\lambda_c \approx 2.4 \times 10^{-12} \text{ m}$

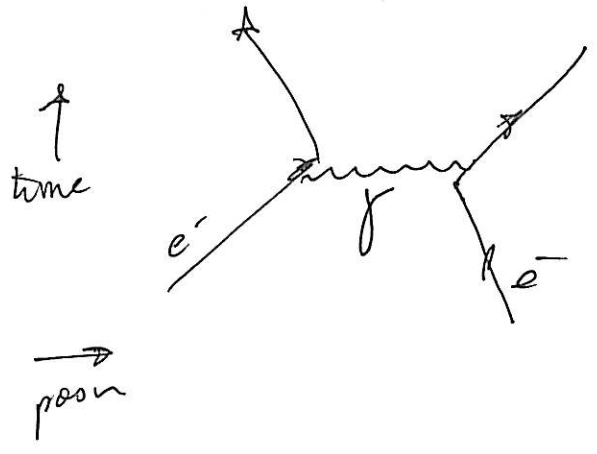
- frequency \downarrow , $\lambda \uparrow$... photon loses energy
- explicit appearance of h !
 \Rightarrow goes away in "classical limit"
 intrinsic to quantum hypothesis
- observation of Compton scattering = stunning confirmation of photon model! (and rel.)
- only for SMALL λ ... X-rays, gammas

Better

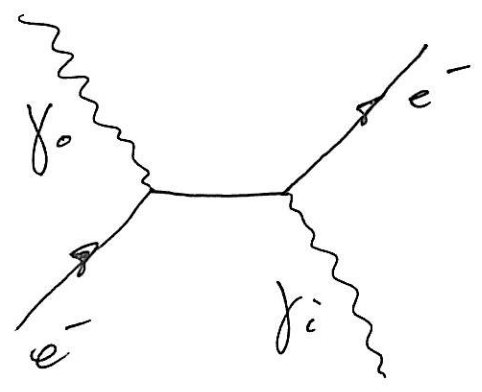
2-step process

- ① absorb photon of $E_i = hf_i, p_i = h/\lambda_i$
- ② emit photon of $E_o = hf_o, p_o = h/\lambda_o$

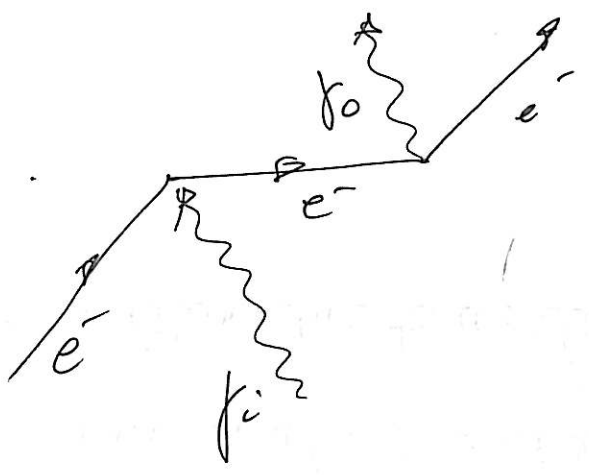
Actually: all interactions of charged particles can be thought of as photon interaction!



- one e^- emits γ , recoils
- 2nd absorbs later, scatters!
- equiv to COULOMB!



Compton: e^- emits photon, recoils, absorbs 2nd one
(time ordering observer-dep)



~~Compton~~
 e^- absorbs photon, recoils, emits 2nd photon