- exam I back tomorrow
- HW4 = solve the other 2 exam problems
  - due tonight
- short quiz tomorrow
  - magnetic forces, fields from wires



magnetism is not like electricity exactly.

magnets come only in +/- or N/S pairs



































almost like a bar magnet (cases where they behave differently are oddballs)



#### Want more? Superposition.

Many loops = many little B's working together. "solenoid"



#### practical application: speakers



back to torque ...



loop *plane* parallel to B - torque loop *plane* perpendicular to B - no force or torque

or ... magnetic moment (perp to area) aligns with field



dc motor : current loop wants to rotate in field tries to align area perpendicular along field

either oscillate current, or push it hard enough to "flip"



## linear motor - current pushes bar out (rail gun)



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## coils of wire or bar magnets - want to align with field



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e- orbiting a nucleus has an orbital momentum L and a magnetic moment due to the orbit.



e- have negative charge -- current is opposite to V, and L is opposite the magnetic moment

# another type of moment - spin

- rough analogue is a spinning ball of charge
- looks like a tiny circulating current

- really, intrinsic angular momentum of particles
- more significant than orbital effect, but moment behavior is qualitatively similar

# origin of magnetic moments?

- Bohr model & orbits is not right, but a start
- orbital motion of H atom = current loop
  dipole moment generated = Bohr magneton

$$\vec{\mu}_B = -\frac{ge}{2m_e}\vec{L} \approx 9.27 \times 10^{-24} \,\mathrm{A/m^2}$$

spin of electron provides another magnetic moment

reality: magnetism does come from L

QM picture

moment comes from angular momentum

orbital momentum is important ...

... but there is also a spin angular momentum

Most materials of interest for us:

orbital moments are negligible

spin moments dominate

# magnetism in real materials

- like dielectrics are electrically polarizable ...
- ... some materials are *magnetically* polarizable
- can be positive or negative

• B inside material is not the same as outside

# Types of Magnetism

- Paramagnetism
  - In gases and molecular liquids and solids the individual ions, atoms, or molecules may have magnetic moments
  - the size of these moments ~ Bohr magneton
  - magnetic energy of one of these molecular magnets in a field of 1T is  $\sim \cos \theta$  60ueV
  - compare to thermal energy of 25meV at RT
  - relatively weak magnetization field compared to the applied field, (M<<H) except at very low temperature</li>

moments try to align with the field - your fridge  $\mu_B \approx 60 \,\mu {\rm eV/T}$   $k_B \approx 86 \,\mu {\rm eV/T}$ 

# Types of Magnetism

- Diamagnetism
  - no spin magnetic moment on an atom or molecule, orbital response dominates
  - usually gives a magnetization opposite to the applied field.
  - extreme case of diamagnetism = superconductor.
    - type I superconductor diamagnetic magnetization exactly cancels the applied field.

## moments try to align opposite in the field

degree of alignment per applied B = susceptibility

negative for diamagnets positive for paramagnets



- Classical Argument gives correct formula
- Some of steps are suspect
- All materials show diamagnetism, often masked by other magnetic responses
- T independent
- All electrons contribute
- Some elements give *positive* susceptibility ... what is this?
- Paramagnetism

# Types of Magnetism

- Ferromagnetism
  - This type of magnetism is observed in the following metals:
    - Transition Metals Fe, Co, Ni
    - Lanthanides (rare earth) Gd, Tb, Dy, Ho, Er, Tm
    - Many alloys and compounds (some of which contain none of these atoms)

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- Ferromagnetism arises from the exchange interaction
- quantum mechanical manifestation of the Coulomb interaction.
- electrons can reduce their Coulomb energy by staying away from each other & increasing their spin angular momentum.
- The exchange interaction is relatively short range.
  - mostly neighboring atoms only
- positive interaction = neighbors parallel = FM



# total moment = diamagnetic GaAs (huge volume) + ferromagnetic Co (small volume)



## magnetic field amplifier - i.e., an electromagnet





how does this relate to NMR/MRI?

what happens to a moment in B?

magnetic field exerts a torque on the moment this causes precession about the field



electron orbit in the yz plane sets up an L which, due to the Lorentz force, rotates in the xy plane

#### moments in a field:

energy change  $\Delta U = -\vec{\mu} \cdot \vec{B}$ 

minimizing U means parallel to field  $\vec{\tau} = \vec{\mu} \times \vec{B}$ 

 $\overrightarrow{\omega}_{1}$  $\stackrel{\rightarrow}{\mu}$ B Ì dø

torque

making torque zero means parallel to field

interaction with field leads to torque ...

torque leads to precession

$$\Rightarrow \vec{\omega}_L = -\gamma \vec{B} = \frac{ge}{2m_e} \vec{B} \approx 14 \,\mathrm{GHz/T}$$

# so what?

- local value of B depends on environment
- so precession frequency depends on environment
  - electron density, electronegativity, induction
- local environment is a function of bonding

 if you can measure the precession frequency, you can ID atom + environment

• (cannot really explain this without quantum)





H atoms in diff environments give diff frequencies

## ratio of peaks ... ratio of number of atoms



http://www.cis.rit.edu/htbooks/nmr/inside.htm



there is current through  $S_1$ , but not through  $S_2$ .

violates Ampere's law?

e- orbiting a nucleus has an orbital momentum L and a magnetic moment due to the orbit.



e- have negative charge -- current is opposite to V, and L is opposite the magnetic moment